

## Note

X-ray single crystal structure analyses of  
5-deoxy-5-*C*-(alkylphosphinyl)-glucopyranoseTatsuo Oshikawa<sup>a\*</sup>, Mitsuji Yamashita<sup>a</sup>, Kuniaki Seo<sup>b</sup>,  
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Received 15 September 1997; accepted in revised form 5 January 1998

**Abstract**

X-ray crystallographic analyses were performed on single crystals of 1,2,3,4-tetra-*O*-acetyl-5-deoxy-5-*C*-(isopropylphosphinyl)- $\beta$ -D-ribofuranoses, 1,2,3,4-tetra-*O*-acetyl-5-deoxy-5-*C*-(isopropylphosphinyl)- $\beta$ -D-xylofuranoses, and 1,2,4-tri-*O*-acetyl-5-deoxy-3-*O*-methyl-5-*C*-(phenylphosphinyl)- $\beta$ -D-ribofuranoses. These compounds have the (*R*<sub>P</sub>) configuration at the phosphorus atom with the <sup>4</sup>C<sub>1</sub> conformation of the pyranose ring, and the conformation supported by NMR studies. © 1998 Elsevier Science Ltd. All rights reserved

**Keywords:** Phospha sugar; NMR; X-ray; Configuration; Conformation

In previous papers [1–4], we reported the syntheses of phospha sugars which were assigned as 1,2,3,4-tetra-*O*-acetyl-5-deoxy-5-*C*-(isopropylphosphinyl)- $\beta$ , $\alpha$ -D-ribofuranoses **5a** and **5b**, 1,2,3,4-tetra-*O*-acetyl-5-deoxy-5-*C*-(isopropylphosphinyl)- $\beta$ , $\alpha$ -D-xylofuranoses **7a** and **7b**, and 1,2,4-tri-*O*-acetyl-5-deoxy-3-*O*-methyl-5-*C*-(*R*, *S*)-(phenylphosphinyl)- $\beta$ -D-ribofuranoses **9a** and **9b**, however, the conformations of these compounds were not determined. Yamamoto et al had previously reported [5–9] the syntheses of phospha sugars, determination of the configuration at phosphorus in the six membered pyranoid rings, and the conformation of the heterocycles by <sup>1</sup>H NMR. The X-ray analyses

of phospha sugar analogs have also been reported by Luger et al [10–14]. We herein report the analyses of 5-deoxy-5-(alkylphosphinyl)-glucopyranoses **5**, **7**, and **9** by <sup>1</sup>H NMR (270 MHz) and X-ray crystallography.

**1. Results and discussion**

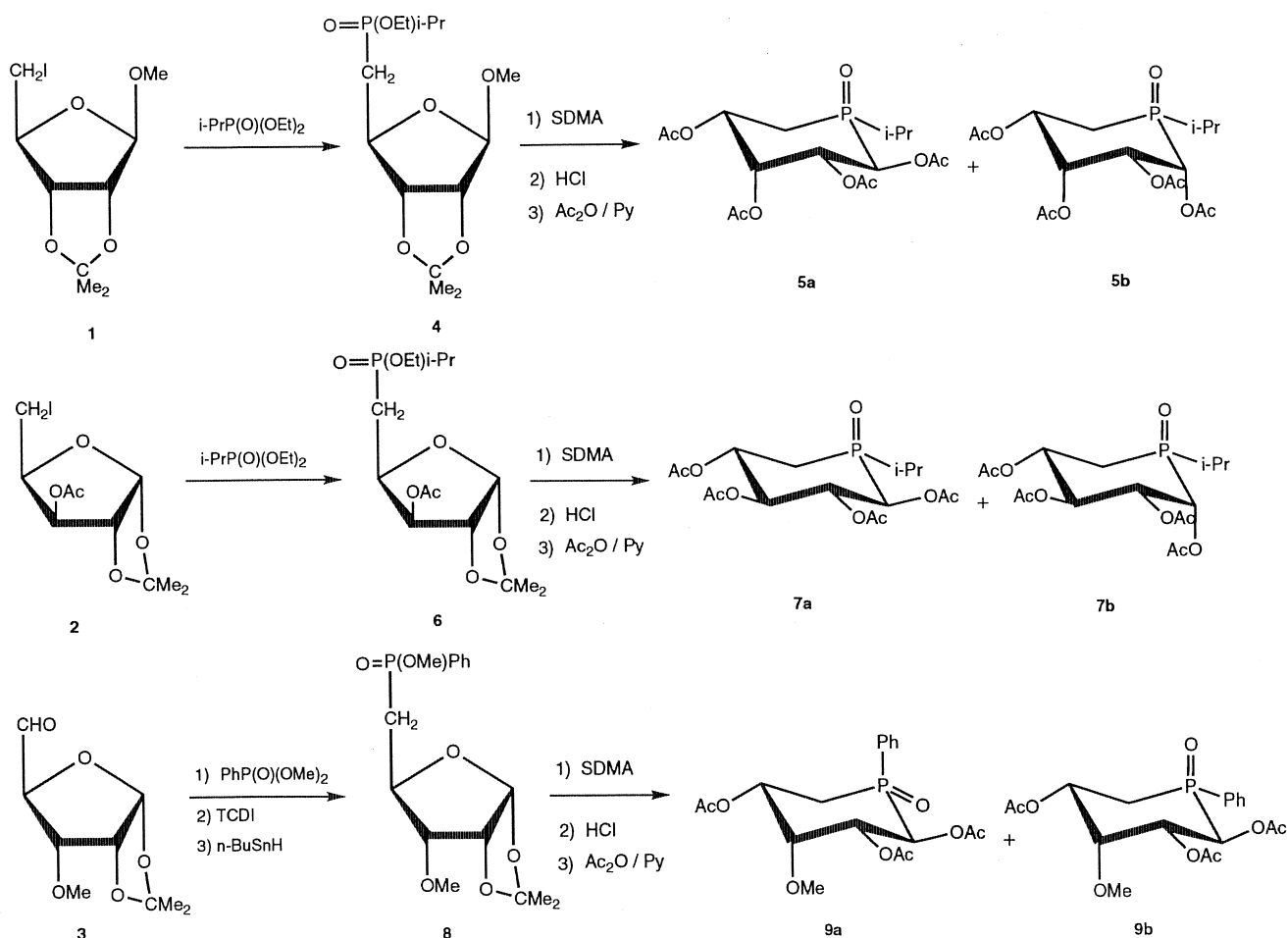
Methyl 5-deoxy-5-*C*-(ethoxyisopropylphosphinyl)-2,3-*O*-isopropylidene- $\beta$ -D-ribofuranoside **4**, 3-*O*-acetyl-5-deoxy-5-*C*-(ethoxyisopropylphosphinyl)-1,2-*O*-isopropylidene- $\alpha$ -D-xylofuranose **6**, and 5-deoxy-1,2-*O*-isopropylidene-5-*C*-(methoxyphenylphosphinyl)-3-*O*-methyl- $\alpha$ -D-ribofuranose **8** were synthesized from methyl 5-deoxy-5-iodo-2,3-*O*-isopropylidene- $\beta$ -D-ribofuranose **1**, 3-*O*-acetyl-5-

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deoxy-5-iodo-1,2-*O*-isopropylidene- $\alpha$ -D-xylofuranose **1** and 1,2-*O*-isopropylidene-3-*O*-methyl- $\alpha$ -D-ribo-pentadialdo-1,4-furanose **2**, respectively, in good yield. Treatment of compounds **4**, **6**, and **8** with sodium dihydrobis(2-methoxyethoxy)aluminate (SDMA) giving 5-deoxy-5-alkyl- or 5-deoxy-5-phenyl-phosphino derivatives followed by ring enlargement of these derivatives by the ring opening and closing two steps known procedure gave pyranose type phospho sugars **5**, **7**, and **9** in good yields (Scheme 1).

Determination of the absolute configuration at the phosphorus atom and the conformational analysis of the pyranoid ring were carried out for phospho sugars **5a**, **5b**, **7a**, **7b**, **9a**, and **9b** prepared previously [1–4]. The precise structures of these compounds were determined on the basis of the 270 MHz  $^1\text{H}$  NMR spectra. The assignments of all signals were readily made employing first-order analysis with the aid of a decoupling technique. The results are summarized in Table 1.  $^4\text{C}_1$  (D)

conformation for all pyranose type phospho sugar derivatives was established. In the case of sugar analogs **5a**, **5b**, **7a**, and **7b** the assignments were difficult, because the signals of Ha-5 and He-5 overlapped with acetyl groups, and H-1, H-2, and H-4 overlapped each other on the  $^1\text{H}$  NMR spectra in  $\text{CDCl}_3$ . When the  $^1\text{H}$  NMR measurements of **5a**, **5b**, **7a**, and **7b** were carried out in benzene- $d_6$ , the signals of H-1, H-2, H-4, H-5a, H-5e, and acetyl protons were separated from each other due to anisotropy of the solvent. In order to assign the  $^1\text{H}$  NMR signals of phospho sugars **5a**, **5b**, **7a**, and **7b**, the chemical shift differences of each proton signal and the dependence of various  $J$  values reflecting the dihedral angles on solvents used were carefully taken into consideration. The  $^1\text{H}$  NMR spectral data of **5a**, **5b**, **7a**, and **7b** generally agree with the characteristic features of  $\delta$  and  $J$  values for pyranose type phospho sugars occurring in an approximate  $^4\text{C}_1$  (D) conformation (see Table 1). Namely, the quasi-axial orientation of the  $\text{P}=\text{O}$



Scheme 1.

Table 1

<sup>1</sup>H NMR chemical shifts and coupling constants for phospho sugars **5**, **7**<sup>a</sup>, and **9**<sup>b</sup>

Diastereomer	Chemical shift ( $\delta$ , ppm)																
	H-1	H-2	H-3	H-4	H-5e	H-5a	AcO-1,2,3,4 <sup>c</sup>				P-CH	P-C-Me	P-C-Me'				
<b>5a</b>	5.23	5.72	5.43	5.50	2.56	1.79	1.69	1.68	1.67	1.65	1.26	0.74	0.89				
<b>5b</b>	5.67	5.64	5.44	5.38	1.79	1.64	1.50	1.45	1.42	1.39	1.26	0.71	0.65				
<b>7a</b>	5.50	5.66	5.74	5.37	1.85	1.61	1.48	1.46	1.41	1.37	1.42 <sup>d</sup>	0.75	0.67				
<b>7b</b>	5.75	5.59	5.41	5.46	2.14	1.33 <sup>c</sup>	1.50	1.47	1.45	1.36	1.10	0.58	0.65				
	H-1	H-2	H-3	H-4	He-5	Ha-5	MeO	AcO-1,2,4 <sup>d</sup>				Ph(o)	Ph(m)	Ph(p)			
<b>9a</b>	6.27	4.83	3.83	4.96	2.87	2.82	3.63	2.06	2.12	2.13	7.75	7.51	7.60				
<b>9b</b>	5.96	5.54	4.06	5.60	2.62	2.41	3.64	1.95	2.08	2.09	7.89	7.57	7.64				
	$J_{1,2}$	$J_{1,P}$	$J_{2,3}$	$J_{2,P}$	$J_{3,4}$	$J_{4,5e}$	$J_{4,5a}$	$J_{4,P}$	$J_{5a,5a}$	$J_{5e,P}$	$J_{5a,P}$	$J_{5e,1}$	$J_{P-CH}$	$J_{CH-Me}$	$J_{P-C-Me}$		
<b>5a</b>	10.5	10.0	4.5	2.5	2.2	4.5	10.3	4.8	14.0	22.9	11.8	0	15.6	7.3	17.8		
<b>5b</b>	2.7	14.0	3.1	0.5	3.0	4.5	12.0	2.2	14.0	23.4	13.6	2.5	14.0	7.0	12.4		
<b>7a</b>	11.1	5.0	9.2	2.2	9.5	4.6	11.5	4.8	14.6	22.9	11.8	0	14.1	7.0	17.0		
<b>7b</b>	2.7	14.4	10.5	0.5	10.1	4.6	12.6	2.2	14.3	23.4	13.6	2.2	14.3	7.3	14.6		
	$J_{1,2}$	$J_{1,P}$	$J_{2,3}$	$J_{2,P}$	$J_{3,4}$	$J_{4,5e}$	$J_{4,5a}$	$J_{4,P}$	$J_{5e,5a}$	$J_{5e,P}$	$J_{5a,P}$	$J_{P,o}$	$J_{P,m}$	$J_{P,p}$	$J_{o,m}$	$J_{m,p}$	$J_{o,p}$
<b>9a</b>	11.6	11.9	3.2	5.8	2.4	3.8	10.5	5.4	15.1	18.4	17.5	11.8	3.3	1.7	7.7	7.7	1.5
<b>9b</b>	11.4	11.6	3.6	3.5	2.4	3.8	13.8	2.2	14.0	15.2	5.4	12.5	3.3	1.7	7.5	7.5	1.5

<sup>a</sup> Determined in C<sub>6</sub>D<sub>6</sub>.<sup>b</sup> Determined in CDCl<sub>3</sub>.<sup>c</sup> The assignments of acetoxyl groups may be interconvertible.<sup>d</sup> Approximation or uncertainty caused by overlapping of acetoxyl signals by may be included.

Table 2

Crystal and structure refinement for **5b**, **7b**, and **9b**

Molecular formula	<b>5b</b>	<b>7b</b>	<b>9b</b>
	C <sub>16</sub> H <sub>25</sub> O <sub>9</sub> P	C <sub>16</sub> H <sub>25</sub> O <sub>9</sub> P	C <sub>18</sub> H <sub>23</sub> O <sub>8</sub> P
Molecular weight	392.34	392.34	398.35
Temperature (K)	293	293	293
Crystal system	Monoclinic	Orthorhombic	Orthorhombic
Space group	P2 <sub>1</sub>	P2 <sub>1</sub> 2 <sub>1</sub> 2 <sub>1</sub>	P2 <sub>1</sub> 2 <sub>1</sub> 2 <sub>1</sub>
Unit cell dimensions (Å)			
<i>a</i>	8.8359	12.435	10.022
<i>b</i>	9.180	16.412	22.337
<i>c</i>	12.6117	10.105	8.796
$\beta$ (°)	94.473		
Volume (Å <sup>3</sup> )	1019.8	2062.2	1969.0
Z (molecules/cell)	4	4	4
Density (calculated, g cm <sup>-3</sup> )	1.555	1.264	1.344
Absorption coefficient (mm <sup>-1</sup> )			
<i>F</i> (000)	832.0	832.0	840.0
Crystal size (mm)	0.2×0.2×0.3	0.2×0.2×0.3	0.2×0.2×0.3
$\theta$ range for data collection (°)	51.1–57.0	48.5–55.0	42.3–54.1
Index ranges for data collection			
Reflections collected	1750	1784	1720
Independent reflections	1633(R <sub>int</sub> = 0.059)		
Refinement method	Full-matrix least-squares on F <sup>2</sup>		
Goodness of fit indicator	2.77	1.54	2.07
Final <i>R</i> indices [ <i>I</i> > 2 $\sigma$ ( <i>I</i> )]			
<i>R</i> 1	0.043	0.035	0.047
<i>wR</i> 2	0.036	0.031	0.038
Largest diff. peak and hole (e Å <sup>-3</sup> )	0.15 and -0.29	0.12 and -0.11	0.25 and -0.20

group in all of these phospho sugar derivatives are established by the downfield chemical shifts of the H-2 and H-4 signals compared with previous reports [9]. The  $\alpha$ -orientation of C-1 is derived by considering the small magnitudes of  $J_{1,2}$  (2.7–3.8 Hz) and  $J_{1,5e}$  (2.2–2.5 Hz) values of **5b** and **7b**,

whereas the  $\beta$ -anomers **5a** and **7a** show large  $J_{1,2}$  (10.0–11.1 Hz) and negligible  $J_{1,5e}$  values. The coupling constants of  $J_{2,3}$  and  $J_{3,4}$  of compounds **5** and **9** showed small Hz (2.2–4.5 Hz), and  $J_{2,3}$  and  $J_{3,4}$  of compounds **7** showed large Hz (9.2–10.5 Hz).

Table 3  
Selected torsion angles for **5b**, **7b**, and **9b**

Sequence	<b>5b</b>	<b>7b</b>	<b>9b</b>
	Angle (degrees)	Angle (degrees)	Angle (degrees)
C-1–C-2–C-3–C-4	–58.0	–58.1	–68.4
C-2–C-3–C-4–C-5	–57.5	59.2	64.1
P-1–C-5–C-4–C-3	–60.8	–62.8	–61.6
C-1–P-1–C-5–C-4	57.1	57.0	55.4
P-1–C-1–C-2–C-3	59.7	57.9	66.6
O-1–P-1–C-6–C-7	–60.7	–57.9	–31.8
O-1–P-1–C-5–C-4	–60.8	–60.2	–65.7
O-1–P-1–C-6–C-8	–58.4	65.2	146.6 <sup>a</sup>
O-4–C-3–C-4–O-5	–68.0	61.3	62.3
O-5–C-4–C-3–C-2	175.6	–179.1	–177.4
O-3–C-2–C-3–O-4	63.3	–63.8	–64.0
O-2–C-1–C-2–O-3	60.6	62.9	–56.4
O-2–C-1–P-1–C-6	49.8	–46.8	74.3

<sup>a</sup> The torsion angle 146.6 degree is that of O-1–P-1–C-6–C-11.

Table 4  
Bond lengths (Å) for compound **5b**, **7b**, and **9b**

<b>5b</b>		<b>7b</b>		<b>9b</b>	
Atoms	Bond lengths	Atoms	Bond lengths	Atoms	Bond lengths
P-1–O-1	1.499(4)	P-1–O-1	1.479(4)	P-1–O-1	1.486(5)
P-1–C-5	1.813(7)	P-1–C-5	1.801(6)	P-1–C-5	1.815(7)
O-2–C-1	1.444(7)	O-2–C-1	1.437(6)	O-2–C-1	1.443(7)
O-3–C-2	1.449(8)	O-3–C-2	1.443(6)	O-3–C-2	1.454(8)
O-4–C-3	1.455(7)	O-4–C-3	1.435(6)	O-4–C-3	1.416(8)
O-5–C-4	1.429(7)	O-5–C-4	1.443(5)	O-5–C-4	1.430(8)
O-6–C-11	1.211(8)	O-6–C-9	1.187(6)	O-6–C-12	1.19(1)
O-10–C-9	1.173(9)	O-8–C-12	1.175(6)	O-27–C-15	1.204(9)
C-1–C-2	1.504(9)	C-1–C-2	1.512(6)	C-6–C-7	1.381(1)
C-2–C-3	1.510(8)	C-2–C-3	1.514(6)	C-7–C-8	1.37(1)
C-3–C-4	1.524(9)	C-3–C-4	1.523(7)	C-8–C-9	1.41(1)
C-4–C-5	1.523(9)	C-4–C-5	1.509(8)	C-9–C-10	1.37(1)
C-6–C-7	1.51(1)	C-6–C-7	1.516(1)	C-10–C-11	1.40(1)
P-1–C-1	1.849(7)	C-10–C-11	1.483(8)	C-15–C-16	1.49(1)
P-1–C-6	1.806(7)	C-12–C-16	1.507(9)	C-17–C-18	1.51(1)
O-2–C-9	1.389(8)	P-1–C-1	1.828(5)	P-1–C-1	1.834(7)
O-3–C-11	1.334(8)	P-1–C-6	1.816(7)	P-1–C-6	1.808(7)
O-4–C-13	1.349(8)	O-2–C-9	1.355(6)	O-2–C-17	1.365(8)
O-5–C-15	1.321(7)	O-3–C-12	1.337(6)	O-3–C-15	1.349(8)
O-9–C-15	1.207(7)	O-4–C-10	1.355(6)	O-4–C-14	1.424(1)
O-11–C-13	1.207(9)	O-5–C-13	1.324(7)	O-5–C-12	1.350(9)
C-6–C-8	1.53(1)	O-7–C-10	1.185(6)	O-7–C-17	1.189(8)
C-9–C-10	1.52(1)	O-9–C-13	1.188(7)	C-1–C-2	1.527(8)
C-11–C-12	1.48(1)	C-6–C-8	1.509(1)	C-2–C-3	1.527(9)
C-13–C-14	1.50(1)	C-9–C-15	1.493(9)	C-3–C-4	1.498(1)
C-15–C-16	1.46(1)	C-13–C-14	1.486(9)	C-4–C-5	1.574(1)
C6–C11	1.395(9)				
C12–C13	1.48(1)				

These tentative stereochemical conclusions promoted us to carry out X-ray crystallographic analyses of **5a**, **7b**, and **9b**. Rod-shaped crystals of **5a**, **7b**, and **9b** were grown from ethyl acetate-hexane. Precise lattice constants and three dimensional intensity data were obtained by a RIGAKU AFC7R controlled Stoe four-circle diffractometer with Ni-filtered  $\text{CuK}\alpha$  radiation. A summary of the

crystallographic data, selected torsion angles, and bond lengths are shown in Tables 2–4, respectively. Phase determination was made by a direct method (SHELXS) [15] and expanded using Fourier techniques [16]. ORTEP plots for compounds **5a**, **7b**, and **9b** are shown by Figs 1–3, respectively. As indicated by the ring torsion angles given in Table 3, the pyranoid rings of **5a**, **7b**, and **9b** have a

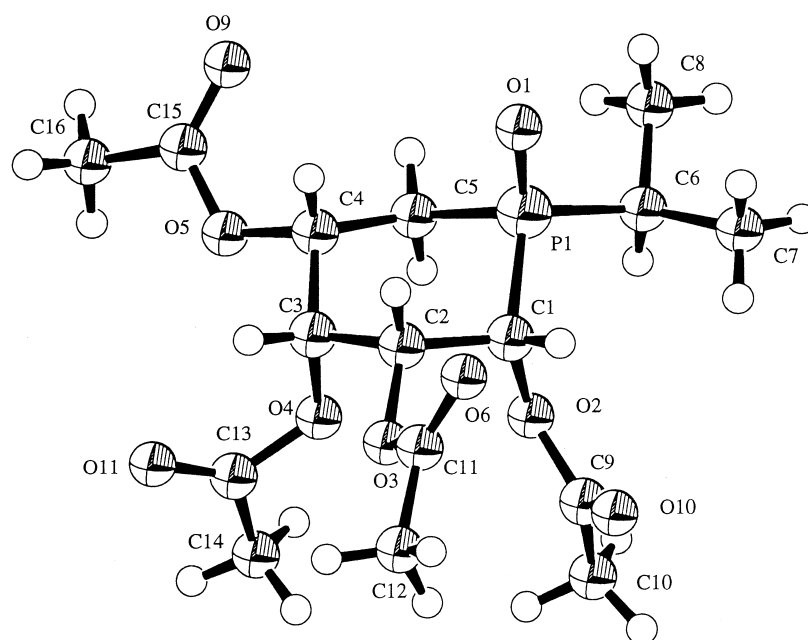


Fig. 1. ORTEP plot for compound **5b**.

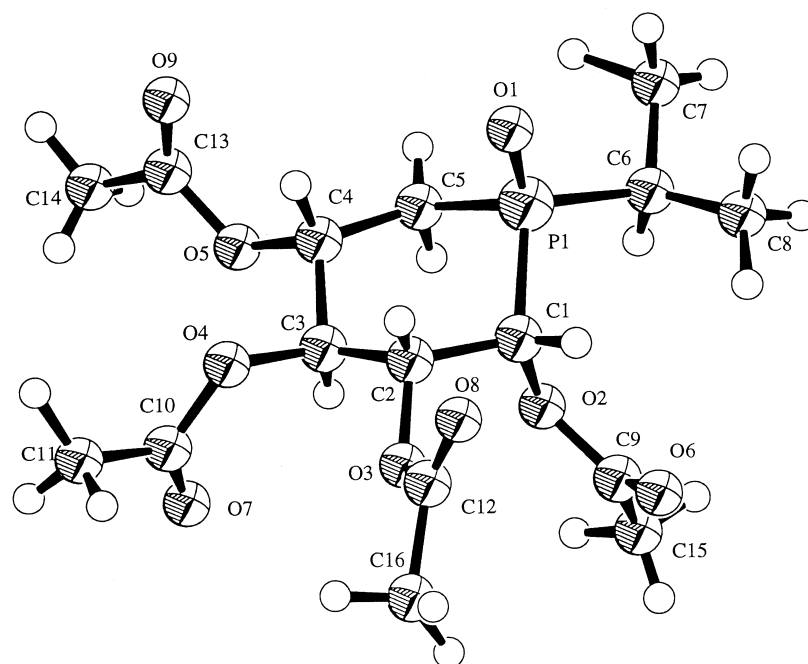
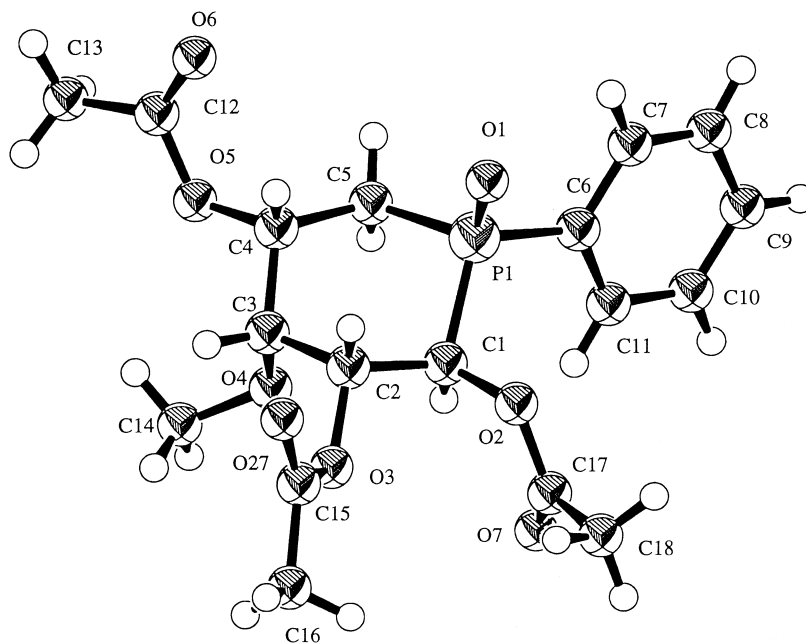


Fig. 2. ORTEP plot for compound **7b**.

Fig. 3. ORTEP plot for compound **9b**.

rather regular  ${}^4C_1$  conformation. As Figs 1–3 show, compounds **5b** and **7b** are 1,2,3,4-tetra-*O*-acetyl-5-deoxy-5-*C*-[(*R*)-isopropylphosphinyl]- $\beta$ -D-ribofuranose and 1,2,3,4-tetra-*O*-acetyl-5-deoxy-5-*C*-[(*R*)-isopropylphosphinyl]- $\beta$ -D-xylofuranose.

These two pyranoses are in  ${}^4C_1$  (D) conformation. In compound **7b**, the substituents at C2, C3, C4, and P link equatorially, while that at C-1 links axially. In compounds **5b** and **9b**, the substituents at C-2, C-4, and P link equatorially, and that at C-3 links axially. In these molecules, the acetoxy groups on C-1 to C-4 are in quasi-equatorial positions. The acetoxy groups on C-1 to C-4 have usual syn-parallel arrangement of the C=O bond with the C–H bond of the adjacent ring atom, and anti-parallel arrangement of the P=O bond with the C–H bond of isopropyl group. On comparison of the present internal torsion angles of six membered ring of the sugar analogs **5a**, **7b**, and **9b** observed by the X-ray crystallographic analyses with those of previous X-ray data for 5-deoxy-5-phosphinyl-D-glucopyranose derivatives, the present torsion angles of C-1–C-2–C-3–C-4 and P-1–C-1–C-2–C-3 are smaller than those of previous structures by ca. 2–7°, and the present torsion angles of P-1–C-5–C-4–C-3 and C-1–P-1–C-5–C-4 are larger than those of previous structures by ca. 2–3° [11–14]. The Cremer–Pople [17] puckering parameters of  $Q=0.646\text{\AA}$ ,  $\theta=9.778^\circ$ ,  $\Psi=161.6^\circ$  for **5b**,  $Q=0.634\text{\AA}$ ,  $\theta=10.6^\circ$ ,  $\Psi=10.0^\circ$  for **7b**, and  $Q=0.693\text{\AA}$ ,  $\theta=5.00^\circ$ ,  $\Psi=27.9^\circ$  for **9b**, respec-

tively. The deviation of C-1 and C-4 from the plane defined by C-2, C-3, C-5 and P are 0.769 and  $-0.689\text{\AA}$  for **5b**, 0.783 and  $-0.656\text{\AA}$  for **7b** and 0.732 and  $-0.718\text{\AA}$  for **9b**, respectively.

## 2. Experimental

Synthetic procedures employed in the present paper for all phosphorus containing sugar analogs were similar to those described in the previous papers [1–4].  ${}^1\text{H}$  NMR Spectra were measured in  $\text{CDCl}_3$  (TMS as the internal standard) on a JASCO EX270 (270 MHz) spectrometer. X-Ray single crystallographic measurements were conducted on a RIGAKU AFC7R using cut crystals of size (0.2×0.2×0.3 mm). Phase determination was made by a direct method (SHELXS) [15] and expanded using Fourier techniques [16]. After convergence of all parameters, final *R* values of 3.8% for **5b**, 4.3% for **7b**, and 4.7% for **9a** were obtained, respectively. In the final stage, refinement was made with anisotropic temperature-factors for all non-hydrogen atoms. The methyl hydrogen atoms were left unrefined.

## Acknowledgement

The authors wish to express their thanks to Professor Y. Kobuke's laboratory of Shizuoka

University for providing facilities for measurement of the  $^1\text{H}$  NMR (270 MHz).

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